

# Safety Data Sheet

according to Federal Register / Vol. 77, No. 58 / Monday, March 26, 2012 / Rules and Regulations

Date of issue: 12/12/2013 Version: 1.0

## SECTION 1: Identification of the substance/mixture and of the company/undertaking

#### 1.1. Product identifier

Product form : Mixture

Product name : Ferric Chloride, 0.025% w/v

Product code : LC14300

### 1.2. Relevant identified uses of the substance or mixture and uses advised against

Use of the substance/mixture : For laboratory and manufacturing use only.

#### 1.3. Details of the supplier of the safety data sheet

LabChem Inc

Jackson's Pointe Commerce Park Building 1000, 1010 Jackson's Pointe Court

Zelienople, PA 16063 - USA T 412-826-5230 - F 724-473-0647 info@labchem.com - www.labchem.com

## 1.4. Emergency telephone number

Emergency number : CHEMTREC: 1-800-424-9300 or 011-703-527-3887

: None.

#### **SECTION 2: Hazards identification**

#### 2.1. Classification of the substance or mixture

#### **GHS-US** classification

Not classified

#### 2.2. Label elements

#### **GHS-US** labelling

No labelling applicable

# 2.3. Other hazards

Other hazards not contributing to the

classification

2.4. Unknown acute toxicity (GHS-US)

No data available

# SECTION 3: Composition/information on ingredients

#### 3.1. Substance

Not applicable

Full text of H-phrases: see section 16

#### 3.2. Mixture

Name	Product identifier	%	GHS-US classification
Water	(CAS No) 7732-18-5	99.975	Not classified
Ferric Chloride, Hexahydrate	(CAS No) 10025-77-1	0.025	Acute Tox. 4 (Oral), H302 Skin Corr. 1C, H314 Eye Dam. 1, H318 Aquatic Acute 2, H401

#### **SECTION 4: First aid measures**

# 4.1. Description of first aid measures

First-aid measures general : Never give anything by mouth to an unconscious person. If you feel unwell, seek medical advice

(show the label where possible).

First-aid measures after inhalation : Assure fresh air breathing. Allow the victim to rest.

First-aid measures after skin contact : Remove affected clothing and wash all exposed skin area with mild soap and water, followed by

warm water rinse.

First-aid measures after eye contact : Rinse immediately with plenty of water. Obtain medical attention if pain, blinking or redness

persist.

First-aid measures after ingestion : Rinse mouth. Do NOT induce vomiting. Obtain emergency medical attention.

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#### 4.2. Most important symptoms and effects, both acute and delayed

Symptoms/injuries : Not expected to present a significant hazard under anticipated conditions of normal use.

#### 4.3. Indication of any immediate medical attention and special treatment needed

No additional information available

# **SECTION 5: Firefighting measures**

### 5.1. Extinguishing media

Suitable extinguishing media : Foam. Dry powder. Carbon dioxide. Water spray. Sand.

Unsuitable extinguishing media : Do not use a heavy water stream.

#### 5.2. Special hazards arising from the substance or mixture

No additional information available

#### 5.3. Advice for firefighters

Firefighting instructions : Use water spray or fog for cooling exposed containers. Exercise caution when fighting any

chemical fire. Avoid (reject) fire-fighting water to enter environment.

Protection during firefighting : Do not enter fire area without proper protective equipment, including respiratory protection.

## **SECTION 6: Accidental release measures**

#### 6.1. Personal precautions, protective equipment and emergency procedures

#### 6.1.1. For non-emergency personnel

Protective equipment : Safety glasses. Gloves.

Emergency procedures : Evacuate unnecessary personnel.

#### 6.1.2. For emergency responders

Protective equipment : Equip cleanup crew with proper protection.

Emergency procedures : Ventilate area.

#### 6.2. Environmental precautions

Prevent entry to sewers and public waters. Notify authorities if liquid enters sewers or public waters.

### 6.3. Methods and material for containment and cleaning up

Methods for cleaning up : Soak up spills with inert solids, such as clay or diatomaceous earth as soon as possible. Collect

spillage. Store away from other materials.

# 6.4. Reference to other sections

See Heading 8. Exposure controls and personal protection.

#### SECTION 7: Handling and storage

#### 7.1. Precautions for safe handling

Precautions for safe handling : Wash hands and other exposed areas with mild soap and water before eating, drinking or

smoking and when leaving work. Provide good ventilation in process area to prevent formation of

vapour.

### 7.2. Conditions for safe storage, including any incompatibilities

Storage conditions : Keep container closed when not in use.

Incompatible products : Strong bases. Strong acids.
Incompatible products : Sources of ignition. Direct sunlight.

#### 7.3. Specific end use(s)

No additional information available

# SECTION 8: Exposure controls/personal protection

### 8.1. Control parameters

Ferric Chloride, Hexahydrate	e (10025-77-1)	
USA ACGIH	ACGIH TWA (mg/m³)	1 mg/m³

### 3.2. Exposure controls

Appropriate engineering controls : Emergency eye wash fountains and safety showers should be available in the immediate vicinity

of any potential exposure. Provide adequate general and local exhaust ventilation.

Personal protective equipment : Avoid all unnecessary exposure.

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Hand protection : Wear protective gloves.

Eye protection : Chemical goggles or safety glasses.

Respiratory protection : Wear appropriate mask.

Other information : Do not eat, drink or smoke during use.

# **SECTION 9: Physical and chemical properties**

#### 9.1. Information on basic physical and chemical properties

Physical state : Liquid
Colour : Yellow.
Odour : None.

Odour threshold No data available рΗ : No data available : No data available Relative evaporation rate (butylacetate=1) Melting point No data available Freezing point : No data available Boiling point No data available : No data available Flash point Self ignition temperature : No data available Decomposition temperature No data available Flammability (solid, gas) No data available Vapour pressure No data available Relative vapour density at 20 °C : No data available : No data available Relative density

Density : 1 g/ml

Solubility Soluble in water. Log Pow No data available Log Kow : No data available Viscosity, kinematic No data available No data available Viscosity, dynamic Explosive properties No data available : No data available Oxidising properties Explosive limits No data available

# 9.2. Other information

No additional information available

# **SECTION 10: Stability and reactivity**

#### 10.1. Reactivity

No additional information available

#### 10.2. Chemical stability

Stable under normal conditions.

#### 10.3. Possibility of hazardous reactions

Not established.

#### 10.4. Conditions to avoid

Direct sunlight. Extremely high or low temperatures.

#### 10.5. Incompatible materials

Strong acids. Strong bases.

#### 10.6. Hazardous decomposition products

Hydrogen chloride. iron oxide.

# **SECTION 11: Toxicological information**

# 11.1. Information on toxicological effects

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Acute toxicity : Not classified

Ferric Chloride, Hexahydrate (10025-77-1)

LD50 oral rat 1872 mg/kg (Rat)

 Water (7732-18-5)

 LD50 oral rat
 ≥ 90000 mg/kg

Skin corrosion/irritation : Not classified
Serious eye damage/irritation : Not classified
Respiratory or skin sensitisation : Not classified
Germ cell mutagenicity : Not classified
Carcinogenicity : Not classified
Reproductive toxicity : Not classified
Specific target organ toxicity (single exposure) : Not classified

Specific target organ toxicity (repeated

exposure)

: Not classified

Aspiration hazard : Not classified

Potential Adverse human health effects and

symptoms

: Based on available data, the classification criteria are not met.

# **SECTION 12: Ecological information**

#### 12.1. Toxicity

Ferric Chloride, 0.025% w/v	
LC50 fishes 1	<

Ferric Chloride, Hexahydrate (10025-77-1)		
LC50 fishes 1	26 ppm (96 h; Pisces; Iron)	
EC50 Daphnia 1	9.6 mg/l (48 h; Daphnia magna; Anhydrous form)	
LC50 fish 2	75.6 mg/l (96 h; Gambusia affinis; Anhydrous form)	
EC50 Daphnia 2	296 - 424 mg/l (96 h; Crangon sp.)	
TLM fish 1	1.2 ppm (144 h; Gasterosteidae; Anhydrous form)	

# 12.2. Persistence and degradability

Ferric Chloride, 0.025% w/v	
Persistence and degradability	Not established.

Ferric Chloride, Hexahydrate (10025-77-1)	
Persistence and degradability	Biodegradability: not applicable. Biodegradability in soil: not applicable. No (test)data on mobility of the substance available.
Biochemical oxygen demand (BOD)	Not applicable
Chemical oxygen demand (COD)	Not applicable
ThOD	Not applicable
BOD (% of ThOD)	Not applicable

Water (7732-18-5)	
Persistence and degradability	Not established.

# 12.3. Bioaccumulative potential

Bioaccumulative potential

Ferric Chloride, 0.025% w/v		
Bioaccumulative potential	Not established.	
Ferric Chloride, Hexahydrate (10025-77-1)		
BCF fish 1	<= 100 (Pisces; Anhydrous form)	
Bioaccumulative potential	No bioaccumulation data available.	
Water (7732-18-5)		

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Not established.

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#### 12.4. Mobility in soil

No additional information available

#### 12.5. Other adverse effects

Other information : Avoid release to the environment.

## **SECTION 13: Disposal considerations**

#### 13.1. Waste treatment methods

Waste disposal recommendations : Dispose in a safe manner in accordance with local/national regulations.

Ecology - waste materials : Avoid release to the environment.

# SECTION 14: Transport information

In accordance with DOT

No dangerous good in sense of transport regulations

#### **Additional information**

Other information : No supplementary information available.

**ADR** 

Transport document description

### Transport by sea

No additional information available

#### Air transport

No additional information available

# **SECTION 15: Regulatory information**

## 15.1. US Federal regulations

Ferric Chloride, Hexahydrate (10025-77-1)	
Not listed on the United States TSCA (Toxic Substances Control Act) inventory	
SARA Section 311/312 Hazard Classes	Immediate (acute) health hazard

### Water (7732-18-5)

Listed on the United States TSCA (Toxic Substances Control Act) inventory

## 15.2. International regulations

### **CANADA**

Ferric Chloride, 0.025% w/v		
WHMIS Classification Uncontrolled product according to WHMIS classification criteria		
Ferric Chloride, Hexahydrate (10025-77-1)		
WHMIS Classification	Class E - Corrosive Material	
Water (7732-18-5)		
WHMIS Classification	Uncontrolled product according to WHMIS classification criteria	

# **EU-Regulations**

No additional information available

# Classification according to Regulation (EC) No. 1272/2008 [CLP]

## Classification according to Directive 67/548/EEC or 1999/45/EC

Not classified

## 15.2.2. National regulations

No additional information available

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# 15.3. US State regulations

No additional information available

# **SECTION 16: Other information**

Other information : None.

Full text of H-phrases: see section 16:

or in philaded. dee dedition ite.	
Acute Tox. 4 (Oral)	Acute toxicity (oral), Category 4
Aquatic Acute 2	Hazardous to the aquatic environment — AcuteHazard, Category 2
Eye Dam. 1	Serious eye damage/eye irritation, Category 1
Skin Corr. 1C	Skin corrosion/irritation, Category 1C
H302	Harmful if swallowed
H314	Causes severe skin burns and eye damage
H318	Causes serious eye damage
H401	Toxic to aquatic life

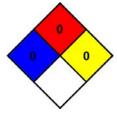
NFPA health hazard : 0 - Exposure under fire conditions would offer no hazard

beyond that of ordinary combustible materials.

NFPA fire hazard : 0 - Materials that will not burn.

NFPA reactivity : 0 - Normally stable, even under fire exposure conditions,

and are not reactive with water.



### **HMIS III Rating**

Health : 0 Minimal Hazard - No significant risk to health

Flammability : 0 Minimal Hazard
Physical : 0 Minimal Hazard

Personal Protection : A

SDS US (GHS HazCom 2012)

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